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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å Disorder in solvent or counterion R factor = 0.035 wR factor = 0.095 Data-to-parameter ratio = 14.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(5R,6S,8S,9R,14R,15R,17R,18S,21S,24R,26S,27R)-5 α -Chloro-16,24-cyclo-13,14-secoergost-2-ene-18,26dioic acid-14:17,14:27-diepoxy-6 β ,13,20,22-tetrahydroxy-1,15-dioxo- γ -lactone δ -lactone methanol solvate monohydrate

The title compound, $C_{28}H_{27}O_{10}Cl\cdot CH_3OH\cdot H_2O$, was isolated from *Physalis minima*. The rigid molecule consists of eight fused rings involving three lactones. The spiro-fused γ -lactone rings are in half-chair and envelope conformations. The spirofused γ -lactone rings are fused to a cyclohexene ring, which is in a half-chair conformation. The outermost cyclohexene ring and δ -lactone rings adopt half-chair and envelope conformations, respectively. Intra- and intermolecular $O-H\cdots O$, C- $H\cdots Cl$ and $C-H\cdots O$ hydrogen bonds are observed. Received 30 August 2005 Accepted 20 September 2005 Online 30 September 2005

Comment

Physalins are the steroidal lactone constituents of *Physalis* and other closely related genera, belonging to the family Solanaceae (Makino, Kawai, Kito *et al.*, 1995). Normal withanolides such as withaferin-A are C-28 steroidal compounds possessing a relatively highly oxidized ergostane-type skeleton and characterized by a six-membered lactone ring in the side chain (Kirson *et al.*, 1971; Glotter *et al.*, 1974). The physalins are biogenetically related to the withanolides and characterized by (*a*) oxidative C13/C14-bond cleavage yielding a nine-membered ring, (*b*) formation of a new six-membered carbocycle between C16 and C24, and (*c*) oxidation of the C13 methyl group to a carboxylic acid, which results in the formation of 18,20-lactonization (Glotter *et al.*, 1974). Physalins are commonly named as 16,24-cyclo-13,14-secosteroids (Makino, Kawai, Kito *et al.*, 1995).



The title compound, (I), known as physalin H, is a steroidal lactone (physalin) previously isolated from *Physalis angulata* (Row *et al.*, 1978; Makino, Kawai, Ogura *et al.*, 1995). Steroidal lactones (withanolides and physalins) exhibit a number of

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organic papers

biological activities, including antitumour (Antoun *et al.*, 1981), immunomodulatory (Budhiraja *et al.*, 1984), antimycobacterial (Pietro *et al.*, 2000), antifungal, anti-inflammatory (Budhiraja *et al.*, 1984, 1986), insect repellant (Glotter, 1991) and immunostimulating (Bates & Eckert, 1972). Compound (I) showed moderate *in vitro* cytotoxic activity against HeLa cells (Makino, Kawai, Ogura *et al.*, 1995). As part of our ongoing research into bioactive constituents from medicinal plants, we have isolated compound (I) from *Physalis minima* for the first time. Fresh plants of *Physalis minima*, locally known as Aknaaj, were collected from Karachi (Pakistan). In the present investigation, we have established the absolute stereochemistry of (I) by single-crystal X-ray diffraction studies (Fig. 1).

Compound (I) is a highly oxygenated steroidal lactone having eight fused rings. The six-membered ring A (C1–C5/ C10) is in a half-chair conformation, with atoms C5 and C10 deviating from the C1/C2/C3/C4 plane by 0.335 (4) and -0.453 (4) Å, respectively. Ring B (C5–C10) is trans-fused to ring A and adopts a chair conformation. The two spiro-fused five-membered rings D (O8/C14-C17) and E (O9/C18/C13/ C17/C20) adopt half-chair and envelope conformations, respectively; in ring D, the deviations of atoms C16 and C17 from the O8/C14/C15 plane are 0.355 (4) and -0.226 (4) Å, respectively, and in ring E atom C17 deviates by 0.638 (3) Å from the O9/C18/C13/C20 plane. Ring F (C16/C17/C20/C22-C24) is in a half-chair conformation, allowing the C24-C16-C17, C16-C17-C20, C17-C20-C22 and C20-C22-C23 bond angles to widen to 114.61 (14), 114.97 (14), 114.43 (14) and 114.83 $(16)^{\circ}$, respectively, while the other internal angles remain close to the tetrahedral value. Ring G (O10/C22-C26) adopts an envelope conformation. The two epoxy seven- and eight-membered rings (O7/C14-C16/C24/C25/C27 and C8/C9/ C11-C14/C17/O8) are in chair and boat-chair conformations, respectively.

The $Csp^3 - Csp^3$ bond distances lie in the range 1.508 (3)– 1.566 (2) Å. The C24–C25–C26 [117.03 (15)°], C9–C11– C12 [119.48 (14)°] and C11–C12–C13 [119.98 (14)°] angles deviate significantly from ideal tetrahedral values, as reported for this class of compounds (Kawai *et al.*, 1970, 1994; Taga *et al.*, 1991). The bond angles involving the spiro atom C17 range from 102.02 (13) to 115.05 (13)°.

The water and methanol molecules of solvation participate in several hydrogen bonds. A number of $O-H\cdots O$, $C-H\cdots Cl$ and $C-H\cdots O$ hydrogen bonds are observed in the molecular and crystal structures. The molecules are linked together by these interactions to form a three-dimensional molecular network (Fig. 2). Most of the intramolecular hydrogen bonds form S(5) or S(6) ring motifs (Bernstein *et al.*, 1995).

Experimental

Air-dried and milled plant material of *Physalis minima* (25.8 kg) was extracted with methanol (120 l) over a period of 21 d at room temperature. After evaporation of the solvent, an extract (2.5 kg) was obtained, which was dissolved in distilled water (5 l) and defatted



Figure 1

The structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. H atoms and methanol and water molecules have been omitted for clarity.





The crystal packing of (I), viewed along the c axis. Dashed lines indicate hydrogen bonds.

with petroleum ether (10 l). The defatted aqueous extract was further fractionated with CH₂Cl₂ (35 l) and *n*-BuOH (5 l), respectively. The CH₂Cl₂ extract was concentrated to a gum (350.5 g) and subjected to column chromatography. A fraction (1.4 g), obtained on elution with CH₂Cl₂–MeOH (95:5), was subjected to repeated column chromatography, which yielded sub-fractions *A* (120.5 mg), *B* (98.8 mg), *C* (50.5 mg), *D* (105.8 mg) and *E* (88.5 mg). Repeated column chromatography on sub-fraction *B* (silica gel 70-230 mesh size) using petroleum ether–acetone (80:20), afforded compound (I) (20 mg), which was recrystallized using CH₂Cl₂–MeOH (70:30) [Physalin H (I): 20 mg, 5.7×10^{-3} % yield, m.p. 548–551 K].

Crystal data

 $C_{28}H_{31}ClO_{10}\cdot CH_4O\cdot H_2O$ $M_r = 613.04$ Orthorhombic, $P2_12_12_1$ a = 13.0931 (5) Å b = 13.5932 (6) Å c = 15.6098 (6) Å V = 2778.19 (19) Å³ Z = 4 $D_x = 1.466$ Mg m⁻³ Mo K α radiation Cell parameters from 15020 reflections $\theta = 2.0-26.5^{\circ}$ $\mu = 0.21 \text{ mm}^{-1}$ T = 293 (2) K Block, colourless $0.54 \times 0.33 \times 0.21 \text{ mm}$ Data collection

Siemens SMART CCD areadetector diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{min} = 0.897, T_{max} = 0.957$ 15825 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.035$ $wR(F^2) = 0.095$ S = 0.855745 reflections 389 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$O11A - H11O \cdots O1W^{i}$	0.96 (4)	1.84 (4)	2.780 (5)	163 (4)
$O1W - H1W1 \cdots O3^{ii}$	0.84	2.08	2.775 (2)	139
$O2-H1O2\cdots O1W^{ii}$	0.82	1.98	2.786 (2)	170
$O4-H1O4\cdots O11A^{ii}$	0.82	1.97	2.718 (5)	152
$O1W - H2W1 \cdot \cdot \cdot O6^{iii}$	0.82	2.03	2.832 (2)	170
$C4-H4B\cdots O4^{ii}$	0.97	2.54	2.898 (3)	102
C7−H7A···Cl1 ⁱⁱ	0.97	2.72	3.1140 (19)	105
$C7-H7B\cdots O7^{ii}$	0.97	2.31	2.696 (2)	102
C9−H9···Cl1 ⁱⁱ	0.98	2.75	3.1814 (17)	107
C9−H9···O2	0.98	2.27	3.065 (2)	138
$C11-H11A\cdots O5^{ii}$	0.97	2.41	3.191 (2)	137
$C11 - H11B \cdot \cdot \cdot O3^{ii}$	0.97	2.38	2.894 (2)	112
$C19-H19B\cdots Cl1^{iv}$	0.96	2.71	3.578 (2)	150
$C19-H19C\cdots O4^{ii}$	0.96	2.38	3.012 (3)	123
$C21 - H21B \cdot \cdot \cdot O2^{ii}$	0.96	2.28	2.871 (2)	119
$C21 - H21C \cdots O8^{ii}$	0.96	2.51	2.871 (2)	102
$C22-H22\cdots O11A^{iii}$	0.98	2.54	3.404 (6)	146

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1$; (ii) x, y, z; (iii) $-x + \frac{3}{2}, -y, z + \frac{1}{2}$; (iv) $x - \frac{1}{2}, -y + \frac{1}{2}, -z + 1$.

The O atom of the methanol molecule is disordered over two positions with occupancies of 0.602 (6) and 0.398 (6). During the refinement, the U^{ij} components of the disordered atoms were restrained to be equal. Atom H11O was located in a difference map and refined isotropically. The water H atoms were located in a difference map and allowed to ride on their parent atom with $U_{iso}(H) = 1.5U_{eq}(O)$ (O–H = 0.82 and 0.84 Å). All other H atoms

5745 independent reflections 5545 reflections with l > 2s(I) $R_{int} = 0.017$ $\theta_{max} = 26.5^{\circ}$ $h = -16 \rightarrow 14$ $k = -17 \rightarrow 16$ $l = -19 \rightarrow 17$

$$\begin{split} &w = 1/[\sigma^2(F_o^2) + (0.0667P)^2 \\ &+ 1.2367P] \\ &where \ P = (F_o^2 + 2F_c^2)/3 \\ (\Delta/\sigma)_{max} = 0.002 \\ \Delta\rho_{max} = 0.25 \ e \ Å^{-3} \\ \Delta\rho_{min} = -0.25 \ e \ Å^{-3} \\ Absolute \ structure: \ Flack \ (1983), \\ &with \ 2520 \ Friedel \ pairs \\ Flack \ parameter: \ 0.02 \ (6) \end{split}$$

were placed in calculated positions (O–H = 0.82 Å and C–H = 0.93– 0.98 Å), with $U_{\rm iso}$ values constrained to be $1.5U_{\rm eq}$ of the carrier atom for the methyl and hydroxyl H atoms, and $1.2U_{\rm eq}$ (C) for the others.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2003).

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References

- Antoun, M. D., Abramson, D., Tyson, L. R., Chang-JER, C., MacLaughlin, J. L., Peck, G. & Carsaday, J. M. J. (1981). J. Nat. Prod. 44, 579–580.
- Bates, R. B. & Eckert, D. J. (1972). J. Am. Chem. Soc. 94, 8258-8260.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.
- Budhiraja, R. D., Garg, K. N., Sudhir, S. & Arora, B. (1986). Planta Med. 52, 28–29.
- Budhiraja, R. D., Sudhir, S. & Garg, K. N. (1984). Planta Med. 50, 134-136.
- Flack, H. D. (1983). Acta Cryst. A**39**, 876–881.
- Glotter, E. (1991). Nat. Prod. Rep. 8, 415-440.
- Glotter, E., Kirson, I., Abraham, A., Sethi, P. D. & Subramanian, S. S. (1974). J. Chem. Soc. Perkin Trans. 1, pp. 1370–1374.
- Kawai, M., Makino, B., Taga, T., Miwa, Y., Yamamoto, T. & Furut, T. (1994). Bull. Chem. Soc. Jpn, 67, 222–226.
- Kawai, M., Matsuura, T. & Osaki, K. (1970). J. Chem. Soc. B, pp. 812-815.
- Kirson, I., Glotter, E. & Lavie, D. (1971). J. Chem. Soc. C, pp. 2032-2044.
- Makino, B., Kawai, M., Kito, K., Yamamura, H. & Butsugan, Y. (1995a). *Tetrahedron*, **51**, 12529–12538.
- Makino, B., Kawai, M., Ogura, T., Nakanishi, M., Yamamura, H. & Butsugan, Y. (1995b). J. Nat. Prod. 58, 1668–1674.
- Pietro, R. C., Kashima, S., Sato, D. N., Januario, A. H. & Franca S. C. (2000). *Phytomedicines*, 7, 335–338.
- Row, L. R., Sarma, N. S., Matsuura, T. & Nakashima, R. (1978). *Phytochemistry*, **17**, 1641–1645.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Taga, T., Miwa, Y. & Machida, K. (1991). Acta Cryst. C47, 2188-2191.

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(5R,6S,8S,9R,14R,15R,17R,18S,21S,24R,26S,27R)-5*a*-Chloro-16,24-cyclo-13,14-secoergost-2-ene-18,26dioic acid-14:17,14:27-diepoxy-6 β ,13,20,22-tetrahydroxy-1,15-dioxo- γ -lactone δ -lactone methanol solvate monohydrate. Corrigendum

Some errors in the paper by Choudhary, Yousuf, Atta-ur-Rahman, Anjum, Fun & Ali [*Acta Cryst.* (2005), E**61**, 03523–03525] are corrected. In the *Abstract*, the text 'The rigid molecule consists of eight fused rings involving three lactones. The spiro-fused γ -lactone rings are in half-chair and envelope conformations. The spiro-fused γ -lactone rings are fused to a cyclohexene ring, which is in a half-chair conformation.' is corrected to 'The rigid molecule consists of eight fused rings involving two lactones. The spiro-fused rings *D* and *E* are in half-chair and envelope conformations. The spiro-fused rings *D* and *E* are in half-chair and envelope conformations. The spiro-fused rings *D* and *E* are in half-chair and envelope conformations. The spiro-fused γ -lactone rings are fused to a cyclohexene ring *F*, which is in a half-chair conformation.' In line 11 of the *Comment*, an atom label is corrected so that the text reads '(*c*) oxidation of the C18 methyl group'.

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